

Spectroscopic Studies of Electronically-Active Defects in Transition Metal Oxides for Advanced Si Devices

Gerald Lucovsky¹ and Jan Lüning²

¹North Carolina State University, Department of Physics, Raleigh, NC 27695-8202, USA

²Stanford Synchrotron Radiation Laboratory, Menlo Park, CA, 94025, USA

Abstract. Based on spectroscopic studies, and guided by ab initio theory, the electron and hole traps in HfO₂ and other transition metal elemental oxides e.g., TiO₂, are assigned to O-atom divacancies, clustered at internal grain boundaries. *Engineering solutions* for defect reduction are identified: i) deposition of ultra-thin, < 2 nm HfO₂ and phase separated Hf silicate dielectrics, in which grain boundary formation is suppressed by effectively eliminating inter-primitive unit cell π -bonding interactions, and ii) non-crystalline Zr/Hf Si oxynitrides without nanocrystalline grain boundaries.

Keywords: transition metal oxides, x-ray absorption spectroscopy, grain-boundary defects, engineering solutions

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INTRODUCTION

Densities of electrically active defects in high-k dielectrics are typically about 10-100 time higher than in Si-SiO₂ devices. A crucial issue is to determine if these defects are intrinsic or if they are derived from processing, e.g., chemical impurities. Spectroscopic studies of high-k gate dielectrics are used to identify electron and hole trapping states in HfO₂ gate stack dielectrics [1]. An asymmetry in hole and electron trapping is a significant limitation for operation and reliability of Si integrated circuits. Based on spectroscopic studies, asymmetries are explained by different charge states of O-divacancies clustered at grain boundaries in nano-crystalline (na-) films. These films are prepared at 30-300°C by remote plasma processing, RPECVD, and reactive evaporation, RE, and processed at higher temperatures (700-900°C) for device integration. Spectroscopic studies have also been performed on na-TiO₂ and -ZrO₂, and complex oxides, e.g., La₂Ti₂O₇, with similar results.

EXPERIMENTAL RESULTS

Soft x-ray photoelectron spectroscopy (SXPS), near edge x-ray absorption spectroscopy (NEXAS), visible and vacuum ultra-violet spectroscopic ellipsometry (vis-VUV SE), and vacuum ultra-violet photoemission spectroscopy (UPS) have been applied to na-Ti, -Zr and -Hf elemental dioxides. Figure 1(a) presents SXPS spectra for HfO₂ and TiO₂, indicating

occupied defect states above the respective valence band edges. Figures 2(a) and (b) compared vis-VUV SE ϵ_2 spectra for TiO₂ and ZrO₂, each displaying features ~1 to 3 eV below the respective d-state band edges [2]. A comparison between the energies of d-state features extracted from the second derivative of spectra of the TiO₂ O K₁ spectrum, with energies of d-state features in the TiO₂ ϵ_2 spectrum yields a linear scaling relationship with a slope of 1 providing the rationale for using d-state energies extracted from O K₁ edges of HfO₂ and ZrO₂, in combination with defect state features in the XPS valence band and VUV SE ϵ_2 spectra, to locate the energies of defect states within the forbidden energy gaps of TiO₂, ZrO₂ and HfO₂. The correspondence between d-state features in O K₁ and ϵ_2 spectra is supported by hole localization in the respective final states: i) a core hole in the O 1s state, and ii) a localized hole in previously doubly occupied valence band edge O-atom 2p π state. The results of this analysis have been used to create electron energy level diagrams for HfO₂ and TiO₂ that are in excellent agreement with molecular orbital (MO) calculations applied to transition metal ion clusters [3]. Spectroscopic data for ZrO₂ yields essentially the same defect states as for HfO₂, with relatively small energy shifts. Photoconductivity and cathode-luminescence data support the intrinsic nature of these defects [2].

Experiments by our group [4], yield essentially the same asymmetric trapping as in Ref. 1. These traps are present in films prepared by RE and RPECVD, and

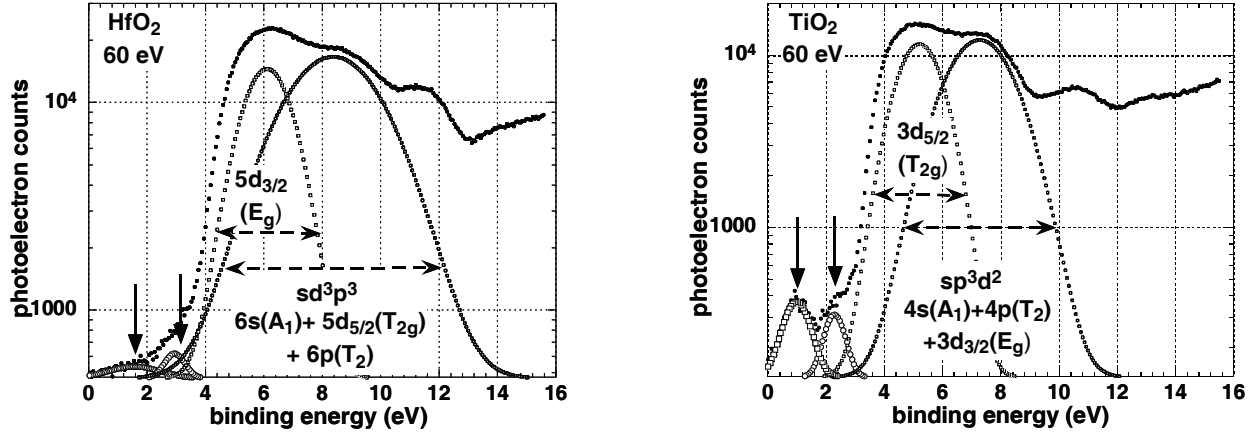


FIGURE 1. SXPS valence band spectra for na-HfO₂ and na-TiO₂, prepared by reactive evaporation.

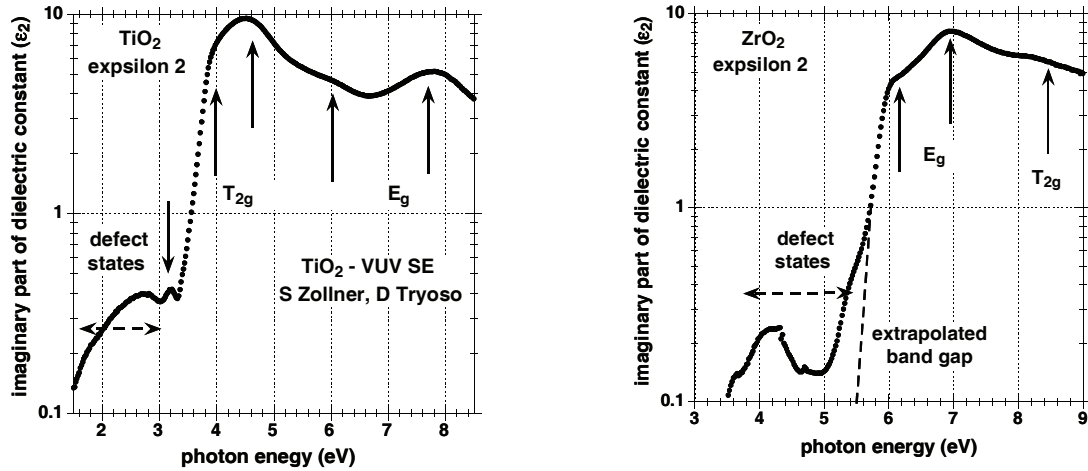


FIGURE 2. ϵ_2 spectra from vis-VUV-SE for (a) na-TiO₂ and (b) na-ZrO₂ films prepared by reactive evaporation.

atomic layer deposition, ALD, as well. Analysis of capacitance-voltage (C-V) and current voltage (J-V) indicate: i) relatively shallow electron traps, ~ 0.5 eV below the conduction band edges of HfO₂ and ZrO₂, as well as ii) deep hole traps below the valence band edge of Si; ~ 3 eV above the HfO₂ valence band.

DISCUSSION

The nature of the intrinsic defects in na-films has been identified by comparing MO calculations for octahedrally coordinated Ti⁴⁺ atoms in TiO₂ with Ti³⁺ in Ti(H₂O)₆³⁺ clusters and Ti₂O₃, as well [3]. The defect electronic structure is *consistent with* the trap being a divacancy. Initial comparisons were based on crystal field splittings, but these have been extended to include E_g and T_{2g} state degeneracy removal by Jahn-Teller (J-T) distortions. The inherent properties of the divacancy defects are: i) electronic states of Ti³⁺ that

fall within the energy gap of the Ti⁴⁺-O states in TiO₂, and J-T degeneracy removal for ii) partially-occupied T_{2g} states at the valence band edge, and iii) unoccupied E_g states at the conduction band edge. The T_{2g} states of Ti³⁺ are partially occupied and act as traps for substrate hole injection, whilst E_g states are empty, and act as traps for transport via substrate injection of electrons [1], and sub-band-gap photoconductivity [2].

Similar energy level schemes apply to i) Zr(Hf)³⁺ defects at grain-boundary clustered divacancies, and ii) Zr(Hf)⁴⁺ states in bulk Zr(Hf)O₂ where the symmetries of the defect d-states are reversed: occupied states near the valence band are J-T term-split E_g states, and those near the conduction band, J-T term split T_g states.

Engineering solutions for elimination of these intrinsic defects have been identified. Two involve suppression of defect states in ultra-thin HfO₂ films HfO₂ films, and chemically-phase separated high HfO₂ content silicates. The third uses the deposition of non-crystalline Si₃N₄-rich Ti(Zr,Hf) Si oxynitrides: e.g.,

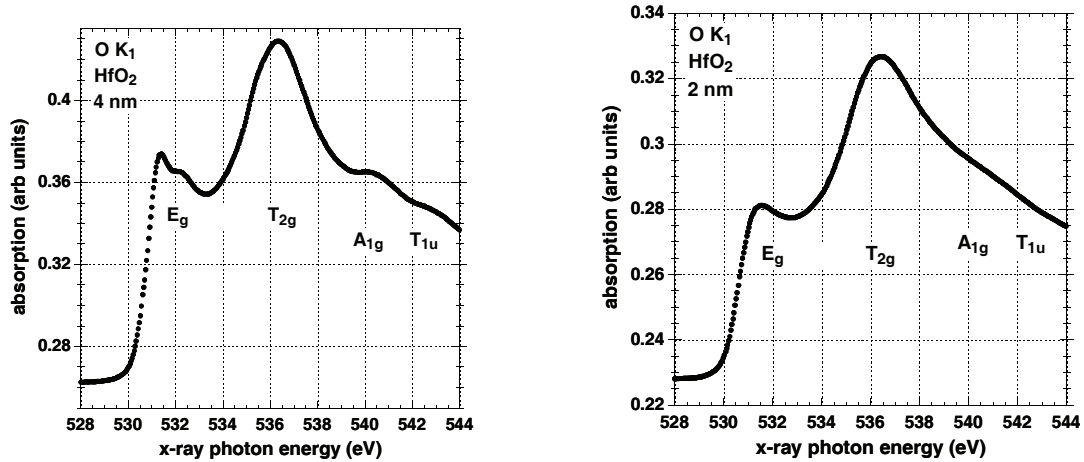


FIGURE 3. O K₁ edge spectra for na-HfO₂: (a) 4 nm thick film, and (b) 2 nm thick film.

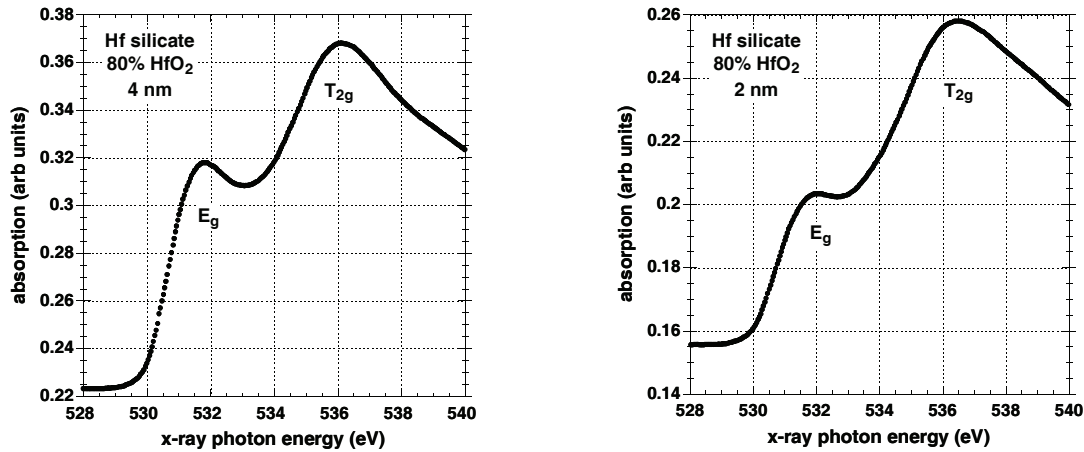


FIGURE 4. O K₁ edge spectra of phase separated Hf silicates (80% HfO₂). (a) 4 nm thick film, and (b) 2 nm thick film.

[Si₃N₄] ~33-40%, with approximately equal [SiO₂] and [Ti(Zr,Hf)O₂] concentrations of 33 to 30%.

In i) ultra-thin [<2 nm] nc-HfO₂, and ii) high HfO₂ (~80%) phase separated Hf silicates, J-T splitting of band edge E_g (5d_{3/2}) anti-bonding π -states is suppressed by decoupling π -bonding interactions between strings of metal and oxide atoms in neighboring *primitive unit cells* (Figs. 3(a) and (b) and 4(a) and (b)). The ~2 nm scale of order is too small for grain boundaries; however, broadening of band edge states occurs, and may effect transport properties.

Studies of O K₁ and N K₁ edges of Zr Si oxynitrides indicate 4-fold coordinated Zr with O nearest-neighbors, and Si, O and N bonding similar to Si₂ON₂. Si₃N₄ rich Zr/Hf Si oxynitrides are stable against phase separation up to 1100°C [5].

Electrical measurements show that ultra-thin HfO₂ annealed in N₂ or NH₃ at 700°C, and Zr/Hf Si

oxynitrides, have low defect densities, and additionally can be scaled to EOT values of 0.7-0.8 nm, thereby extending device scaling below the 0.45 nm node [6].

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